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Refrigerator and Air Conditioning

Basic Concept

VCRS

Ref

VARs

RBC

Ref Equipment

Books: CP Arora

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Psychrometry

Summer & Winter AC

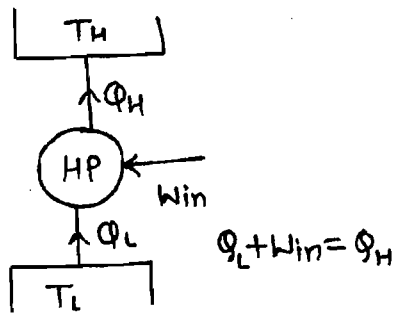
BASIC CONCEPTS

- Refrigeration Effect :- It is the amount of heat which is required to extract from the storage space in order to provide & maintain lower temperature than that of surroundings.

Refrigerant \rightarrow It is the working fluid or working substance which is used to extract the heat from the storage space.

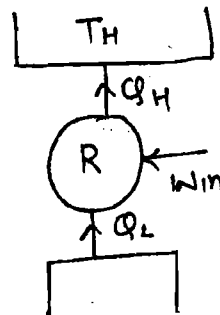
COP \rightarrow Coefficient of Performance or Energy Performance or EPR ratio \rightarrow

$$\text{COP} = \frac{DE}{W_{in}}$$



$$\text{(COP)}_{\text{HP Actual}} = \frac{Q_H}{Q_H - Q_L}$$

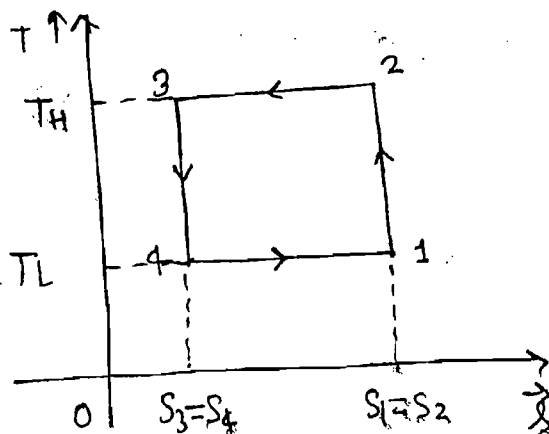
$$\text{(COP)}_{\text{HP Ideal}} = \frac{T_H}{T_H - T_L}$$



$$\text{(COP)}_{\text{R Actual}} = \frac{Q_L}{Q_H - Q_L}$$

$$\text{(COP)}_{\text{R Ideal}} = \frac{T_L}{T_H - T_L}$$

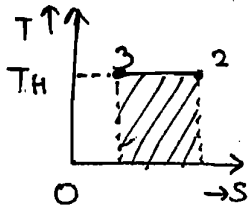
- Ideal Refrigeration Cycle or Reversed Carnot Cycle \rightarrow



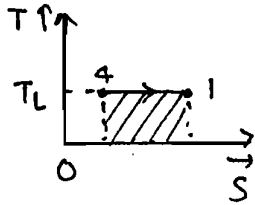
- Process 1-2 Rev. adiabatic Compression
 Process 2-3 Isothermal Heat rejection
 3-4 Isentropic Expansion
 4-1 Isothermal heat addition

$$\text{COP} = \frac{DE}{W_{NET}}$$

$$W_{NET} = Q_{NET} = \cancel{Q_{1-2}} + Q_{2-3} + \cancel{Q_{3-4}} + Q_{4-1}$$



$$dQ_{2-3} = T(S_F - S_I) = T_H(S_3 - S_2) = -T_H(S_1 - S_4) \quad \text{--- (2)}$$



$$dQ_{4-1} = T_L(S_1 - S_4) \quad \text{--- (3)}$$

Use eqⁿ (2) & (3) in eqⁿ (1)

$$W_{NET} = Q_{NET} = -T_H(S_1 - S_4) + T_L(S_1 - S_4)$$

$$W_{NET} = Q_{NET} = (T_L - T_H)(S_1 - S_4) \quad \text{--- (4)}$$

$$W_{NET} = -ive$$

From eqⁿ (4) we can say that our system under consideration is a work absorbing device.

$$W_{input} = (T_H - T_L)(S_1 - S_4)$$

$$\text{COP} = \frac{DE}{W_{input}} = \frac{Q_{4-1}}{(T_H - T_L)(S_1 - S_4)} = \frac{T_L(S_1 - S_4)}{(T_H - T_L)(S_1 - S_4)}$$

$$\text{COP} = \frac{T_L}{T_H - T_L}$$

NOTE:-

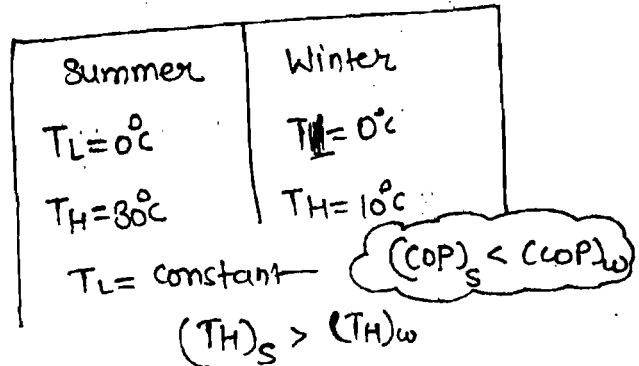
1. Reversed Carnot COP is a function of temp. limits only
2. If there are 'n' number of Rev. Refrigerator are operating between same temp. limits with different working fluids, then the value of max. possible COP or Ideal COP or Reversed Carnot COP are having same value.
3. Reversed Carnot COP is independent of working fluid
4. Producing Ice at 0°C

(a) $(\text{COP})_{\text{summer}} > (\text{COP})_{\text{winter}}$

~~(b)~~ $(\text{COP})_s < (\text{COP})_w$

(c) $(\text{COP})_s = (\text{COP})_w$

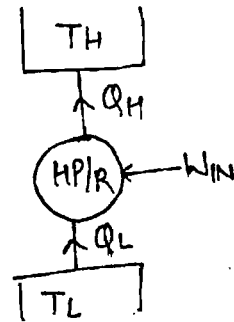
(d) can't say



Relationship between Heat Pump COP & COP of Refrigerator: →

$$\text{COP}_{\text{HP}} = \text{COP}_R + 1$$

$$1 + \text{COP}_R = \frac{T_L}{T_H - T_L} + 1 = \text{COP}_{\text{HP}}$$



The above expression is applicable b/w same temp. limits